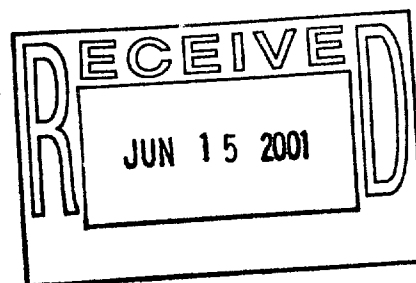


OL010135P

Supplementary Information (De Boos, Fullbrook, Percy)

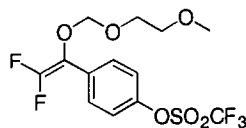
1. Procedures and full data for **4c**, **11e**; ^{19}F and R_f data for all other compounds.
2. GC trace for **4c**.
3. HPLC trace (3 wavelengths) for **11e**.
4. Crude ^{19}F NMR spectra for reaction of **4c** with:
 - a) PhSCI
 - b) NCS
 - c) Br_2
 - d) I_2
 - e) TMSCl/MeOH
5. ^{13}C (a), ^1H (b) and ^{19}F (c) NMR spectra for **4c**.
6. ^{13}C (a), ^1H (b) and ^{19}F (c) NMR spectra for **11e**.
7. Elemental analysis report for **11e**.



Supporting Information:

General Procedure for Stille couplings to stannane 2:

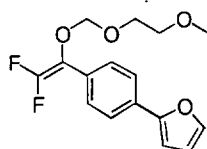
4-[(2,2-Difluoro-1-(2-methoxy-ethoxymethoxy)-vinyl)-phenyl] trifluoromethanesulfonate 4c



A mixture of copper(I) iodide (3.97 mmol, 0.756 g, 22 mol%), triphenylphosphine (1.79 mmol, 0.469 g, 10 mol%) and palladium(II) acetate (0.48 mmol, 0.107 g, 2.7 mol%) was pump-purged twice with argon, then dry, thoroughly degassed DMF (45 ml) was added. 4-Iodophenyl triflate 3c (18.54 mmol, 6.53 g) was added and the mixture heated to 50°C. Stannane 2 (17.76 mmol, 8.12 g) was subsequently added and the mixture stirred for 16 hours at this temperature. The mixture was diluted with diethyl ether and the mixture extracted with water (100 ml) and diethyl ether (3 × 25 ml). The combined organic extracts were dried (MgSO₄) and concentrated *in vacuo*. Purification by column chromatography (Silica gel; 5 % diethyl ether in petroleum ether then 20% ethyl acetate in petroleum ether) afforded triflate 4c (6.32 g, 96 %) as a pale orange oil (97 % by GC); *R*_f 0.44 (20% ethyl acetate in light petroleum); *v*_{max} (film/cm⁻¹) 1731m, 1503m, 1428s, 1271m, 1215s, 1142s, 889s; δ_{H} (300.13 MHz, CDCl₃) 7.50 (2H, d, ³*J*_{HH} 8.88 Hz, CH), 7.24 (2H, d, ³*J*_{HH} 8.88 Hz, CH), 4.82 (2H, s, OCH₂O), 3.80-3.77 (2H, m, OCH₂CH₂O), 3.49-3.46 (2H, m, OCH₂CH₂O), 3.31 (3H, s, OCH₃); δ_{C} (75.47 MHz, CDCl₃) 155.8 (t, ¹*J*_{CF} 292 Hz), 148.9 (t, ⁵*J*_{CF} 2.3 Hz, C_q), 130.7 (dd, ³*J*_{CF} 4.7, 1.7 Hz, C_q), 128.5 (dd, ⁴*J*_{CF} 6.2, 3.4 Hz, CH), 121.6 (s, CH), 118.7 (q, ¹*J*_{CF} 320 Hz), 114.6 (dd, ²*J*_{CF} 34.4, 19.5 Hz), 95.8 (ca. t, ⁵*J*_{CF} 3 Hz), 71.5, 68.7, 59.0; δ_{F} (282 MHz, CDCl₃) -95.79 (d, ²*J*_{FF} 51.5 Hz), -104.51 (d, ²*J*_{FF} 51.5 Hz), -73.0 (s, CF₃); *m/z* (ES, M⁺+Na) 415.1 (100%); [HRMS (ES, M⁺+Na) Found: 415.0247. Calc. For C₁₃H₁₃O₆F₅NaS: 415.0251].

General Procedure for Suzuki couplings to triflates 4b or 4c under conditions a:

2-{4-[2,2-Difluoro-1-(2-methoxy-ethoxymethoxy)-vinyl]-phenyl}-furan 9e

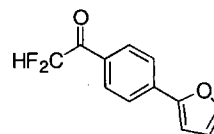


A mixture of dichlorobis(triphenylphosphine)-palladium(II) (0.028 mmol, 0.02 g, 5.5 mol%) and benzenboronic acid (0.96 mmol, 0.12 g) was pump-purged twice with nitrogen,

then taken up in dry, degassed DMF (2 ml). The mixture was warmed to 90°C. Triethylamine (2.0 mmol, 0.28 ml) was added at 30°C, followed by a solution of triflate 4c (0.52 mmol, 0.21 g) in DMF (1 ml). The resulting dark black solution was stirred at 90°C for 90 hours, then diluted with diethyl ether. Water was added to wash out DMF, then the aqueous phase was extracted with diethyl ether (3 × 5 ml). The combined organic extracts were dried (MgSO₄) and concentrated *in vacuo* to afford a crude brown oil. Dissolution with acetone, followed by adsorption onto silica afforded an orange powder. Purification by column chromatography (silica gel; 20% diethyl ether in light petroleum) afforded furan 9e (76 mg, 49%) as a colourless oil; *R*_f 0.19 (20% diethyl ether in light petroleum); δ_{H} (300.13 MHz, CDCl₃) 7.67 (2H, d, ³*J*_{HH} 8.64 Hz), 7.47 (2H, d, ³*J*_{HH} 8.64 Hz), 7.35-7.34 (1H, m), 6.67-6.65 (1H, m), 6.46-6.44 (1H, m), 4.89 (2H, s), 3.87-3.84 (2H, m), 3.56-3.52 (2H, m), 3.36 (3H, s); δ_{F} (282 MHz, CDCl₃) -105.52 (d, ²*J*_{FF} 54.6 Hz), -97.29 (d, 54.6 Hz). This material was cleaved as described below without further characterisation.

General procedure for protic cleavage of MEM ethers 8 or 9:

2,2-Difluoro-1-(4-furan-2-yl-phenyl)-ethanone 11e



Enol acetal 9e (0.87 mmol, 0.27 g) was dissolved in methanol (6 ml) and cooled to 0°C. Chlorotrimethylsilane (1.58 mmol, 0.2 ml) was then slowly added and the resulting solution allowed to warm to ambient temperature, and stirred overnight. Evaporation of the solution afforded crude difluoroketone 11e as a dark green solid. Purification by flash column chromatography (silica gel; 15% diethyl ether in light petroleum) of pre-adsorbed material afforded difluoroketone 11e (0.12 g, 60%) as a white solid (100% by HPLC); mp 113-114°C; *R*_f 0.58 (15 % diethyl ether in petroleum ether); (Found: C, 64.66; H, 3.75%. C₁₂H₈F₂O₂ requires: C, 64.87; H, 3.63%); *v*_{max} (film/cm⁻¹) 1695w, 1608w; δ_{H} (300.13 MHz, CDCl₃) 8.06 (2H, d, 8.5 Hz), 7.87 (2H, d, 8.5 Hz), 7.70-7.69 (1H, m), 7.09-7.08 (1H, m), 6.84 (1H, t, ²*J*_{HF} 53.1 Hz), 6.59-6.57 (1H, m); δ_{C} (75.47 MHz, CD₃COCD₃) 157.4 (s, C_q), 149.5 (s, CH), 141.3 (s, CH), 135.7 (s, C_q), 135.1 (s, CH), 128.8 (s, CH), 117.6 (s, CH), 115.2 (t, ¹*J*_{CF} 248 Hz), 114.1 (s, CH), CO signal absent; δ_{F} (282 MHz, CDCl₃) -121.71 (d, ²*J*_{HF} 47.4 Hz); [HRMS (EI, M⁺) Found: 222.0495. Calc. For C₁₂H₈O₂F₂: 222.0492]; *m/z* (EI, M⁺) 222 (42%), 171 (100%), 143 (17%), 115 (53%).

¹⁹F NMR and R_f data for all compounds:

Compound No.	¹⁹ F NMR data	R _f	Eluent
4a	-106.05 d J=50.8 Hz, -98.40 1H, d J=50.8 Hz, -74.84 (3F) s	0.15	10% DE in LP
4b	-103.09 d J=49.4 Hz, -94.59 1H, d J=49.4 Hz, -72.83 (3F) s	0.27	20% DE in LP
4c	-104.51 d J=51.5 Hz, -95.79 d J=51.5 Hz, -73.0 (3F) s	0.44	20% EA in LP
4d	-103.60 d J=49.6 Hz, -94.18 d J=49.6 Hz, -73.41 (3F) s	0.22	30% DE in LP
4e	-109.91 d J=63.0 Hz, -101.16 d J= 63.0 Hz	0.25	20% DE in LP
4f	-103.78 d J=51.2 Hz, -95.11 d J=51.2 Hz	0.38	40% EA in LP
4g	-99.47 d J=38.2 Hz, -92.28 d J=38.2 Hz	0.22	30% DE in LP
4h	-102.42 d J=51.5 Hz, -98.76 d J=51.5 Hz	0.27	20% DE in LP
5	-119.85 dd J=79.1, 2.5 Hz; -100.73 dd, J=79.0, 16.4 Hz	-	-
6	-102.20 (2F) ca. d J=47.6 Hz, -95.20 (2F) ca. d J=47.6 Hz	-	-
8a	-106.69 d J=55.1 Hz, -98.18 d J=55.1 Hz	0.11	10% DE in LP
8b	-105.37 d J=54.2 Hz, -96.73 d J=54.2 Hz	-	-
8g	-106.03 d J=56.5 Hz, -97.53 d J=56.5 Hz	0.27	20% DE in LP
9a	-105.91 d J=55.2 Hz, -97.48 d J=55.2 Hz	0.41	30% DE in LP
9b	-104.93 d J=53.2 Hz, -96.42 d J=53.2 Hz	-	-
9c	-105.91 d J=55.4 Hz, -97.48 d J=55.4 Hz	0.17	20% DE in LP
9d	-106.68 d J=56.5 Hz, -98.18 d J=56.5 Hz	0.25	20% DE in LP
9e	-105.52 d J=54.6 Hz, -97.29 d J=54.6 Hz	0.19	20% DE in LP
9f	-105.37 d J=53.7 Hz, -97.09 d J= 53.7 Hz	0.21	20% DE in LP
9h	-105.78 d J=53.7 Hz, -97.67 d J=56.5 Hz	0.21	10% DE in LP
10e	-104.78 d J=51.5 Hz, -96.18 d J=51.5 Hz	0.28	20% EA in LP
11a	-122.15 (2F) d J=53.7 Hz	0.55	20% EA in LP
11e	-121.71 (2F) d J=47.4 Hz	0.58	15% DE in LP
11h	-122.00 (2F) d J=50.8 Hz	0.70	15% DE in LP
11g	-122.29 (2F) d J=53.9 Hz	0.55	20% EA in LP
4c + Br ₂	-72.68 (3F) s, -52.56 (2F) s	-	-
4c + I ₂	-72.61 (3F) s, -55.41 (2F) s	-	-
4c + NCS	-72.70 (3F) s, -61.38 (2F) s	-	-
4c + PhSCl	-77.95 (2F) s, -72.67 (3F) s	-	-

DE=diethyl ether

EA=ethyl acetate

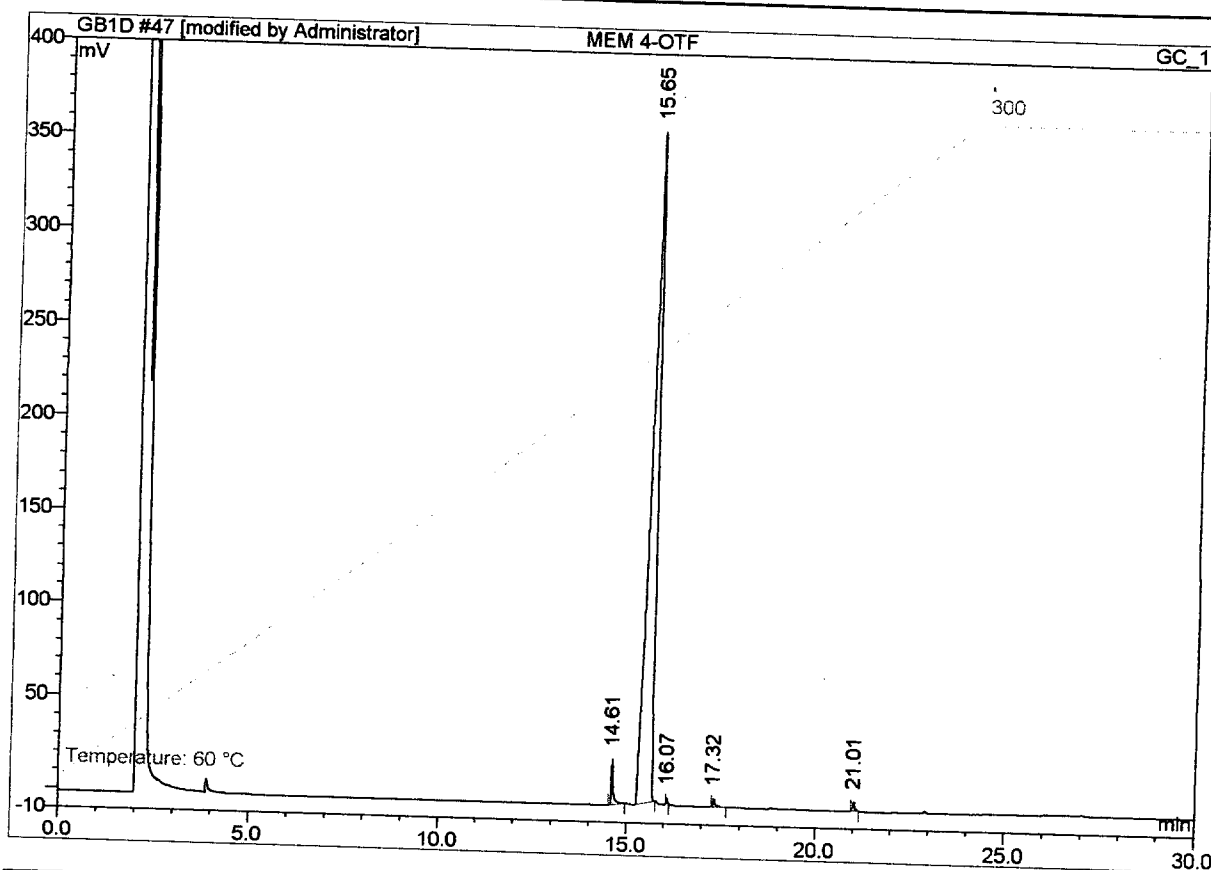
LP=light petroleum (40-60°C)

Unless stated otherwise, all ¹⁹F NMR signals correspond to 1 F atom

47 MEM 4-OTF

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Run Time (min): 30.00

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Bandwidth: n.a.
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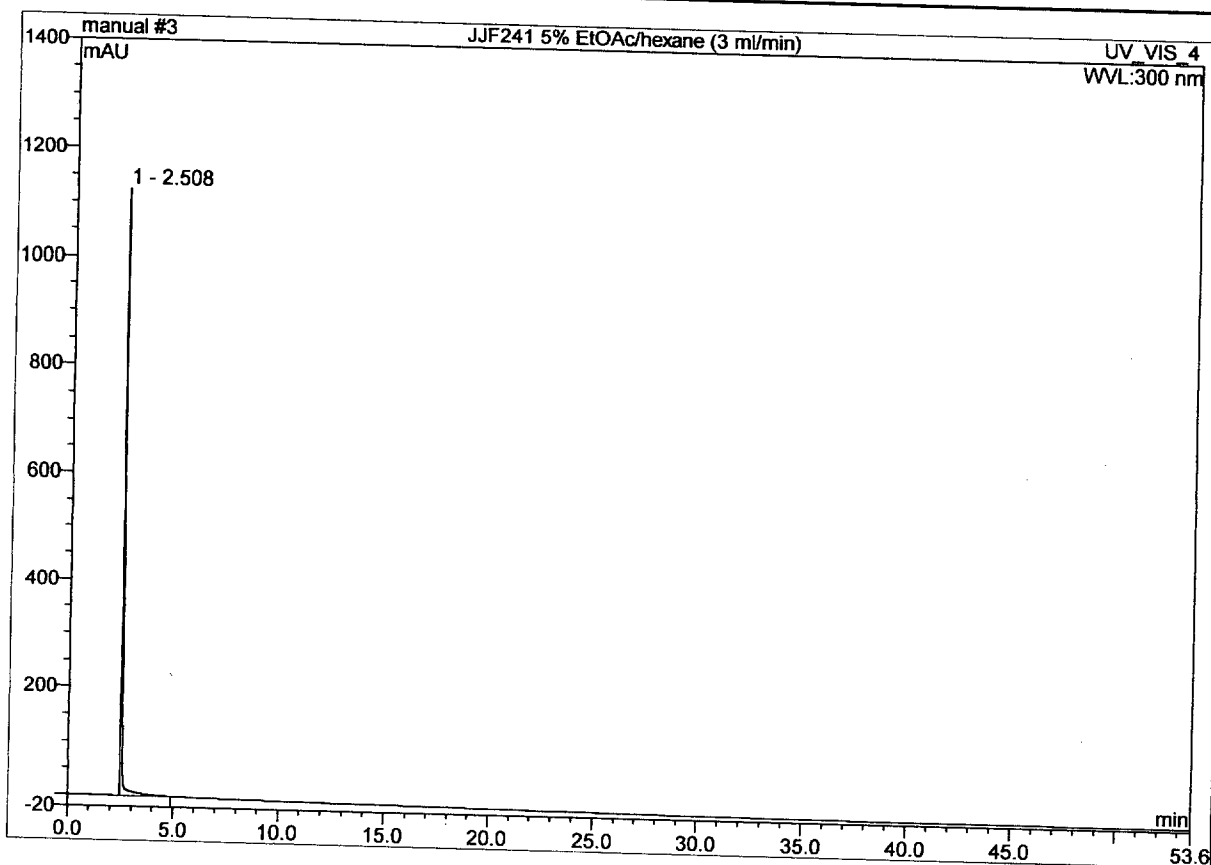


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2	15.65	n.a.	357.685	67.571	96.83	n.a.	BMB
3	16.07	n.a.	3.420	0.137	0.20	n.a.	BMB*
4	17.32	n.a.	4.475	0.319	0.46	n.a.	BMB*
5	21.01	n.a.	4.859	0.292	0.42	n.a.	BMB*
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31/5/01 12:07 PM

3 JJF241 5% EtOAc/hexane (3 ml/min)

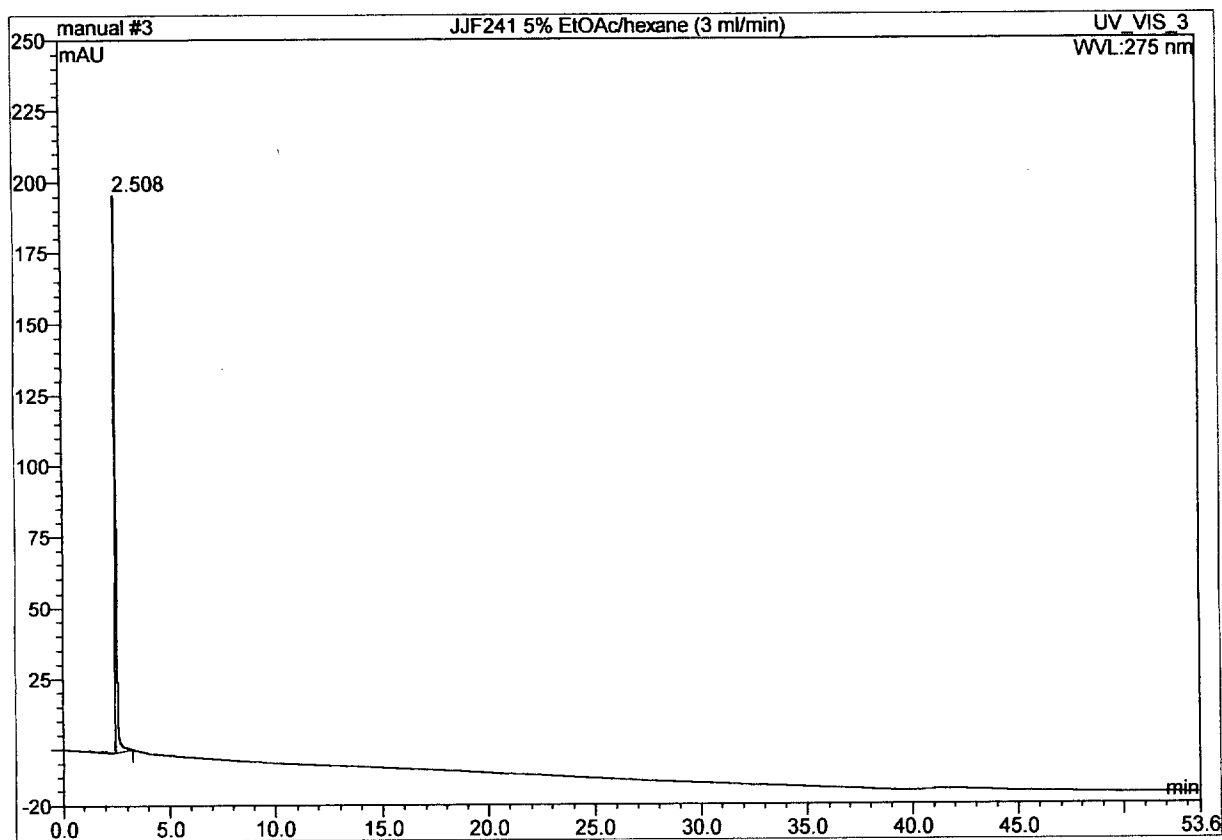
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Quantif. Method:	default	Dilution Factor:	1.0000
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Run Time (min):	53.62	Sample Amount:	1.0000



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3 JJF241 5% EtOAc/hexane (3 ml/min)

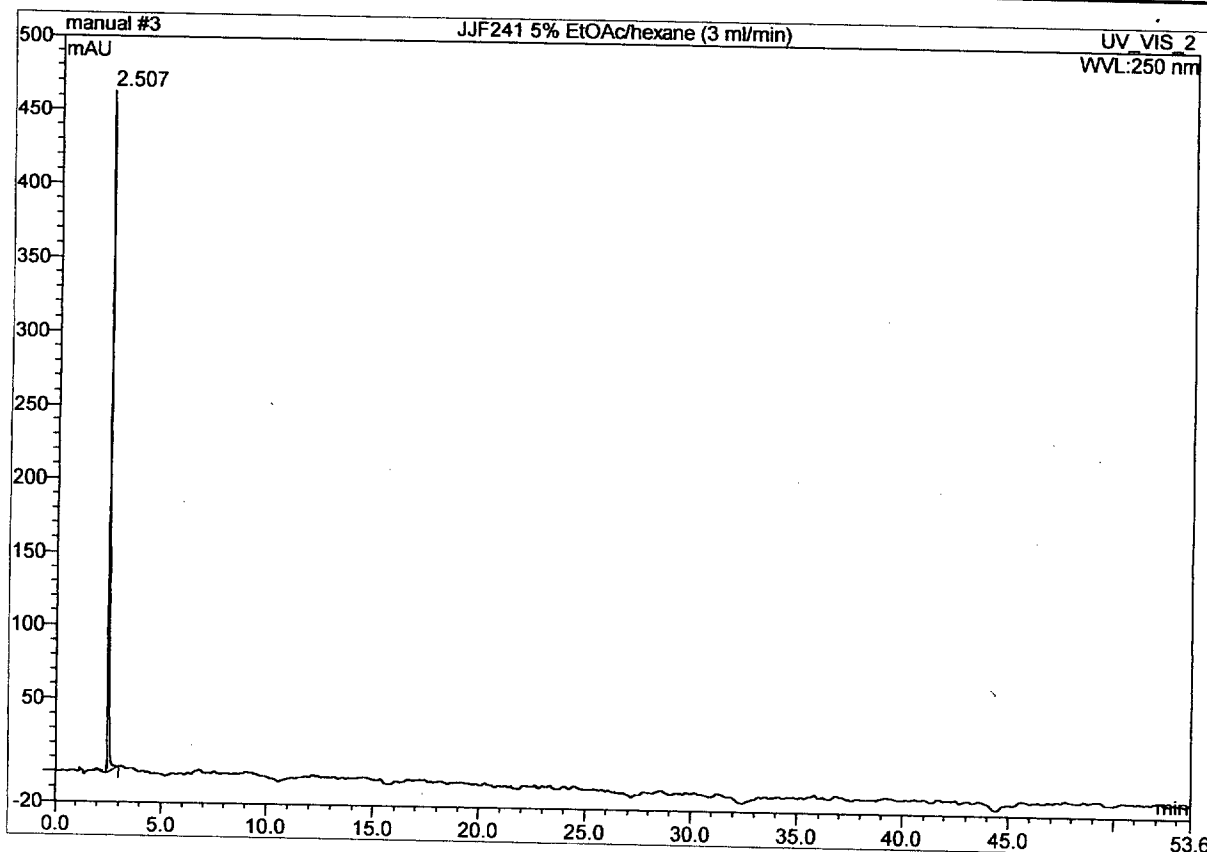
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Quantif. Method:	default	Dilution Factor:	1.0000
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Run Time (min):	53.60	Sample Amount:	1.0000



No.	Ret.Time min	Peak Name	Height mAU	Area mAU*min	Rel.Area %	Amount	Type
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3 JJF241 5% EtOAc/hexane (3 ml/min)

Sample Name:	JJF241 5% EtOAc/hexane (3 ml/min)	Injection Volume:	20.0
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Sample Type:	unknown	Wavelength:	250
Control Program:		Bandwidth:	1
Quantif. Method:	default	Dilution Factor:	1.0000
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3304

ANS - ANDREA DI SALVO



MO20S 1.11

AU PROG

X03 AU

DATE 2-3-73

SA NA 806405

SA NO MR02 1.11

SF 282.407

SY 282.0

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SI 65536

TO 65536

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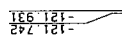
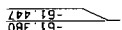
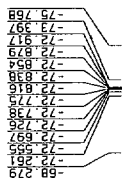
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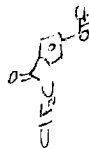
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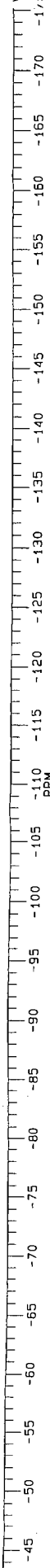
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3

2



52991

ANS - ANDREA DI SALVO

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BMJX R

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DATE 23-2-73

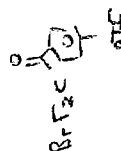
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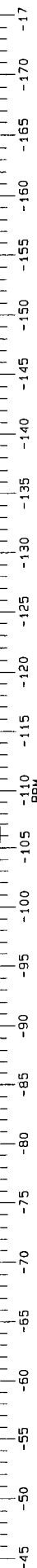
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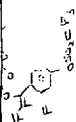
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clean

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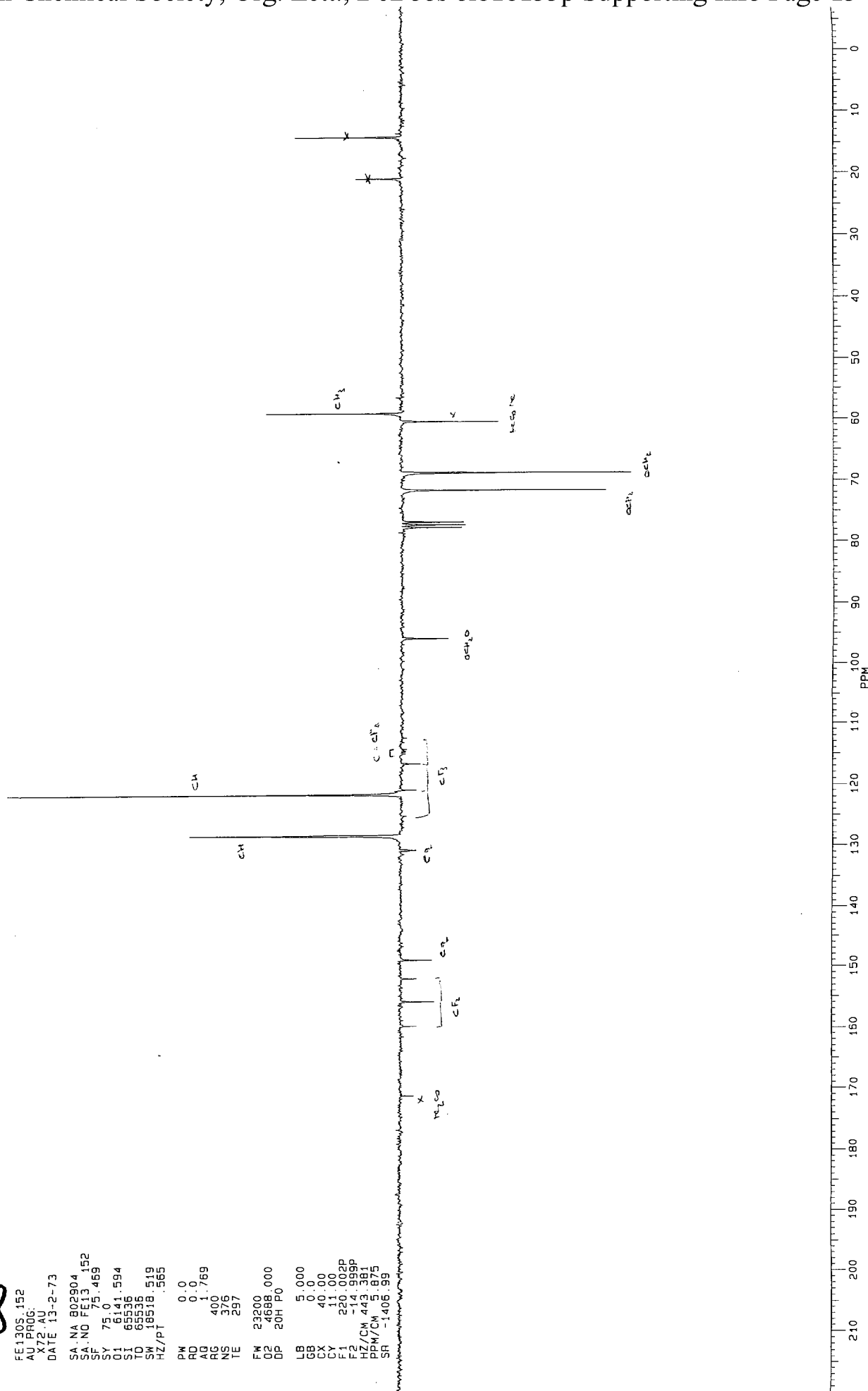




JJF - JEREMY J FULLBROOK

BIOA

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X72 AU
DATE 13-2-73
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SA NO FE13.152
SF 75.469
SY 75.0
O1 6141.594
SI 65536
TD 65536
SW 18518.519
HZ/PT .565
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RD 0.0
AQ 1.769
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CY 11.00
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PPM/CM 5.875
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3134

Jeremy Fullbrook, CDCl₃, +27°C, AV300
¹H, Barcode label 2901

Current Data Parameters

NAME 12-13.43.FAC
 EXPNO 10
 PROCNO 1

F2 - Acquisition Parameters

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 FIDRES 0.182955 Hz
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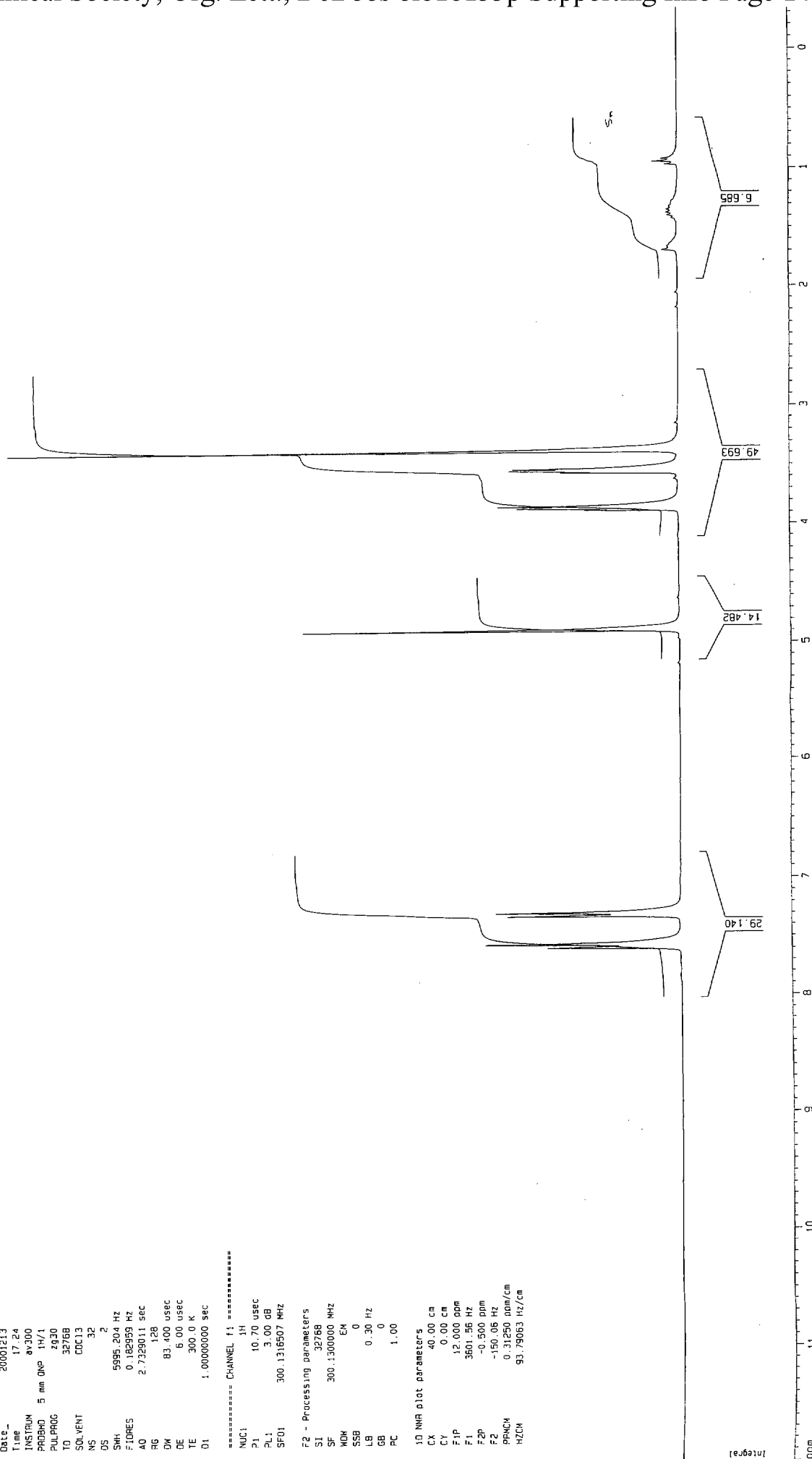
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 F2P -0.500 ppm
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 PRACH 0.31250 ppm/cm
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JJF - JEREMY J FULLBROOK

8/10/01

SP2806.132

AU: PROG:

X10. AU

DATE 28-9-72

SA NA 802807

SA NO SP28.132

SF 282.408

SY 282.0

O1 -5787.000

SI 65536

TD 65536

SW 41666.667

HZ/PT 1.272

PW 0.0

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AG 10.786

NS 32

TE 297

FW 52100

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HZ/CM 988.420

PPM/CM 3.500

SR 27500.39

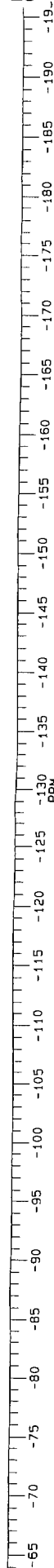
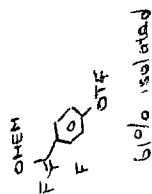
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73.614



LWO - LAURENCE OLCKSIK



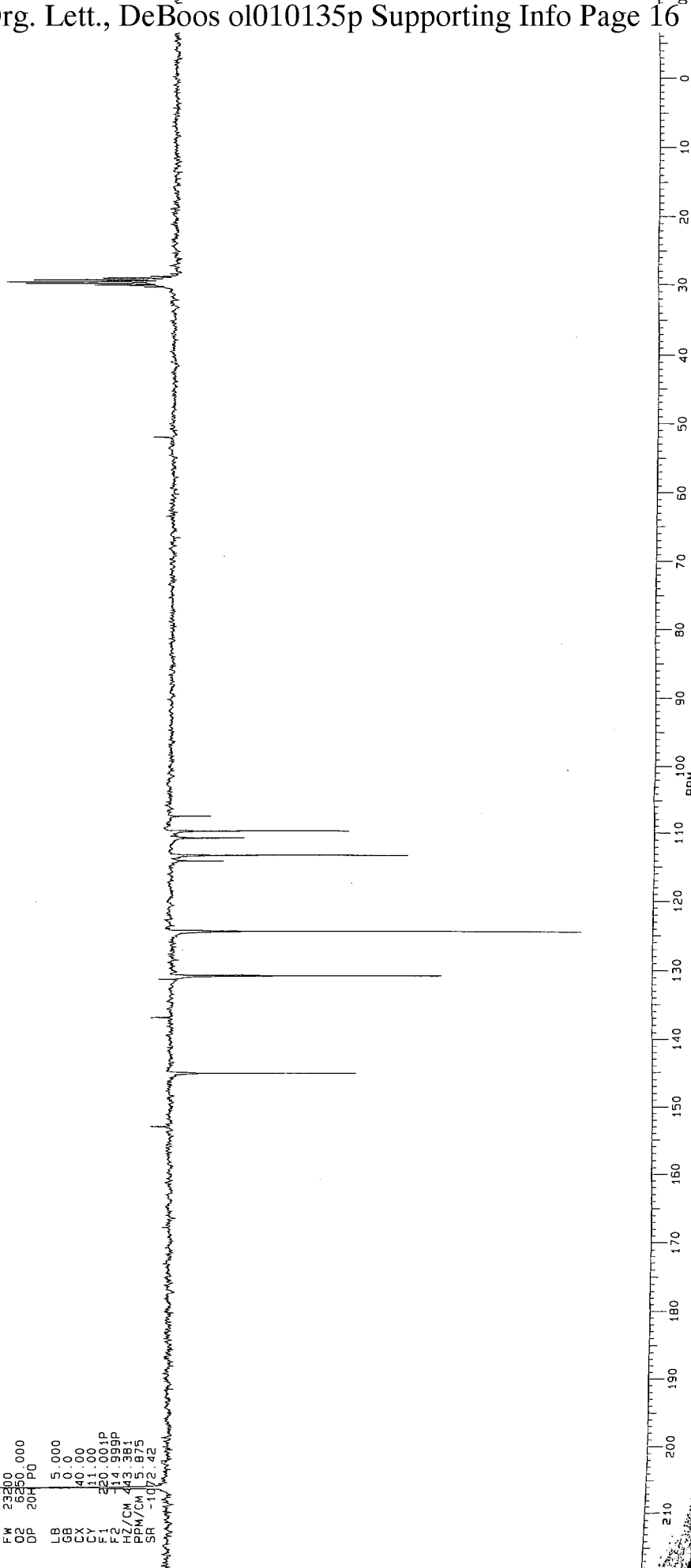
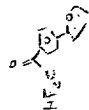
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AU PRG
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DATE 20-4-73

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SI 65536
TD 65536
SW 19518.519
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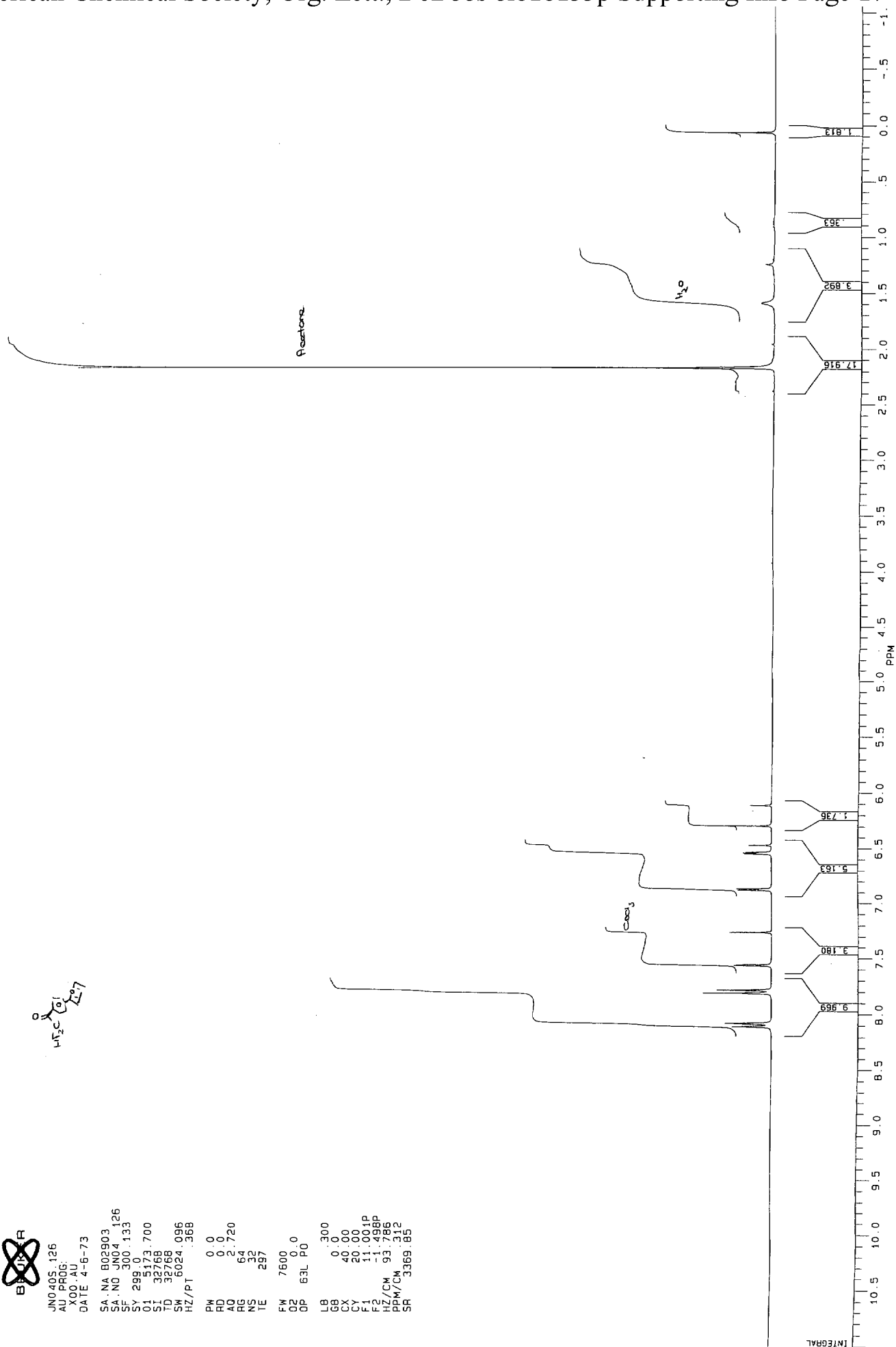
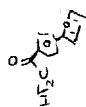
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PPM/CM 5.875
SR -1072.42



3241 JJF - JEREMY J FULLBROOK



JN0405.126
AD PR06:
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DATE 4-6-73
SA NA 802903
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SV 299.0
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CY 20.00
F1 11.001P
F2 -11.498P
HZ/CM 93.785
PPM/CM 312
SR 3369.85



JJF - JEREMY J FULLBROOK

~~SECRET~~

hdd

JN0405.116

AU PRQ:

X03 AU

DATE 4-6-73

SA NA 802913

SF NO 3024.116

SY 282.407

O1 4147.000

SI 65436

TD 65936

SM 41666.667

HZ/PT 1.272

PW 0.0

RD 0.0

AQ 786

RG 64

NS 32

TE 297

FW 52100

O2 4900.000

DP 63L PO

LB 1.000

GB 0.0

CX 40.00

F1 50.007P

F2 -173.886P

HZ/CM 988.420

PEM/CM 3.500

SR 27499.73

-121.560
-121.749

-45 -50 -55 -60 -65 -70 -75 -80 -85 -90 -95 -100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175

Method Filename : MAY2001.MTH

Company name : Birmingham Uni

Operator ID : Lianne Hill

Group No : 25	Element %		
Sample Name	Nitrogen	Carbon	Hydrogen
J241	0.00	64.66	3.75
		64.87	3.63

1 Sample(s) in Group No : 25

Component Name

Nitrogen

Carbon

Hydrogen

I added VO_5 to the sample to aid combustion. This seems to have sorted out the problem so if you have any more sample I will run that so you have 2 sets of results!